

A VERSATILE ROUTE FOR THE SYNTHESIS OF 3-(1',1'-DIMETHYLALLYL)COUMARINS

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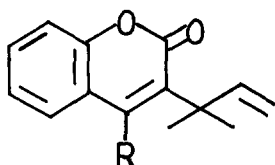
ABSTRACT Synthesis of 3-(1',1'-dimethylallyl)herniarin and 3-(1',1'-dimethylallyl)desmethoxyherniarin has been carried out by a versatile route

During the last few years, a number of coumarins having a 1,1-dimethylallyl unit at position 3 have been isolated from natural sources and many of them have been found to possess hypotensive¹ and spasmolytic^{2,3} activities. The structures of these coumarins have been assigned on the basis of spectral studies only. The only method⁴⁻⁶ recorded in literature for the synthesis of 3-(1',1'-dimethylallyl)coumarins is by the most unusual triple Claisen rearrangement of 7-(3',3'-dimethylallyloxy)coumarin derivatives in which a number of products are formed and yields are very poor (5-14%). In view of this it was considered of interest to develop a convenient method for their synthesis in order to provide synthetic evidence for their structures. In the present communication, we report a facile route for the synthesis of such coumarins in good yields.

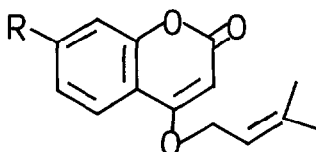
As a test case, the synthesis of 3-(1',1'-dimethylallyl)desmethoxyherniarin (1) has been carried out starting from 4-(3',3'-dimethylallyloxy)coumarin (2). Coumarin 2 was prepared by reaction of 4-hydroxycoumarin with 1-bromo-3-methylbut-2-ene in acetone in presence of anhyd. potassium carbonate, (yield 40%), m.p. 98-99° (lit.⁷ m.p. 98-99°). Claisen rearrangement of 2 in acetic anhydride in presence of sodium acetate (2 hr reflux) afforded 4-acetoxy-3-(1',1'-dimethylallyl)coumarin (3, yield 95%), m.p. 102-103 (lit.⁷ m.p. 99-100°). The structure of 3 was assigned on the basis of its ¹H NMR spectral data which showed, besides other usual signals, an ABX pattern (typical for a vinyl group attached to a quaternary carbon), ¹H NMR: 1.55 (6H, s, CMe₂), 2.34 (3H, s, OAc), 5.00, 5.04 & 6.25 (3H, ABX type, J=17Hz, 10Hz & 1Hz, -CH=CH₂), 7.34 (4H, m, aromatic protons). 3 was directly tosylated with p-toluene sulphonyl chloride in acetone in presence of K₂CO₃ to give 3-(1',1'-dimethylallyl)-4-tosyloxycoumarin (4, yield 50%), m.p. 134-135°. The structure of 4 was in agreement with its ¹H NMR spectral data which showed signals at 1.53 (6H, s, CMe₂), 2.45 (3H, s, CH₃), 4.88, 4.93 & 6.13 (3H, ABX type, J=17Hz, 10Hz & 1Hz, -CH=CH₂), 7.28 (6H, m, aromatic protons), 7.80 (2H, d, J=8.5Hz, aromatic protons). The final step involved the reductive detosylation of 4 with granulated zinc and HCl in alcohol (1½ hr reflux) to give the required coumarin 1 (yield 45%), m.p. 67-68°. It was characterised on the basis of its ¹H NMR spectral data which showed signals at 1.50 (6H, s, CMe₂), 5.03, 5.08 & 6.18 (3H, ABX type, J=17Hz, 10Hz & 1Hz, -CH=CH₂), 7.18-7.43 (4H, m, aromatic protons), 7.53 (1H, s, H₄), IR $\nu_{\text{max}}^{\text{KBr}}$ 1710, 1610, 1450, 1365, 1285 cm⁻¹. This constitutes the first synthesis of 1.

Similarly, 3-(1',1'-dimethylallyl)herniarin (5), a naturally occurring coumarin, has been synthesised as follows. Prenylation of 4-hydroxy-7-methoxycoumarin with 1-bromo-3-methylbut-2-ene in acetone/ K_2CO_3 gave 4-(3',3'-dimethylallyloxy)-7-methoxycoumarin (6, yield 42%), m.p. 93-94°, 1H NMR 1.73 & 1.79 (6H, each s, CH_3), 3.81 (3H, s, OCH_3), 4.65 (2H, d, $J=6.5$ Hz, $-CH_2-$), 5.38 (1H, bt, $J=6.5$ Hz, $-CH=$), 5.73 (1H, s, H_3), 6.78 (2H, m, H_6 & H_8), 7.67 (1H, d, $J=9$ Hz, H_5). On Claisen rearrangement in $Ac_2O/AcONa$ afforded 4-acetoxy-3-(1',1'-dimethylallyl)-7-methoxycoumarin (7, yield 95%), m.p. 77-78° which was assigned the structure on the basis of its 1H NMR spectral data 1.53 (6H, s, CH_3), 2.28 (3H, s, OAc), 3.81 (3H, s, OCH_3), 5.92, 5.97 & 6.24 (3H, ABX type, $J=17$ Hz, 10 Hz & 1 Hz, $-CH=CH_2$), 6.73 (2H, m, H_6 & H_8), 7.17 (1H, d, $J=9.5$ Hz, H_5). Tosylation of 7 with p-toluene sulphonyl chloride in acetone/ K_2CO_3 furnished 3-(1',1'-dimethylallyl)-7-methoxy-4-tosyloxycoumarin (8, yield 55%), m.p. 104-105°, 1H NMR 1.40 (6H, s, CH_3), 2.36 (3H, s, CH_3), 3.67 (3H, s, OCH_3), 4.64, 4.69 & 6.01 (3H, ABX type, $J=17$ Hz, 10 Hz & 1 Hz, $-CH=CH_2$), 6.65 (2H, m, H_6 & H_8), 6.95 (1H, d, $J=9$ Hz, H_5), 7.21 & 7.64 (each 2H, each d, $J=8.5$ Hz, aromatic protons). On reductive detosylation with Zn/HCl in alcohol yielded 5 (yield 40%), m.p. 125-127° (lit.⁸ m.p. 126-128°). Its structure was in agreement with its 1H NMR spectral data which showed signals at 1.55 (6H, s, CH_3), 3.81 (3H, s, OCH_3), 5.06, 5.17 & 6.18 (3H, ABX type, $J=17$ Hz, 10 Hz & 1 Hz, $-CH=CH_2$), 6.68 (2H, m, H_6 & H_8), 7.13 (1H, d, $J=9$ Hz, H_5), 7.48 (1H, s, H_4). IR ν_{max}^{KBr} 1685, 1610, 1355 & 1245 cm^{-1} . Compound 5 was found to be identical with the natural sample.⁸

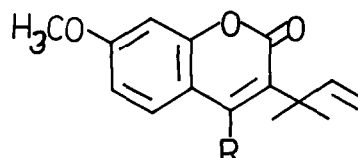
All compounds were analysed well for C and H (within $\pm 0.4\%$). The 1H NMR spectra were taken in $CDCl_3$ using TMS as an internal standard, chemical shifts in δ scale.



(1) R = H



(2) R = H



(3) R = H

(3) R = OAc

(6) R = OCH_3

(7) R = OAc

(4) R = OTs

(8) R = OTs

Acknowledgement We are thankful to the Council of Scientific and Industrial Research, New Delhi for financial assistance.

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(Received in UK 26 February 1982)